

## NOTES.

**Correction.**—Lieut. E. B. Vliet has called our attention to a number of errors in the numbers appearing in the last column (headed “% deviation”) of Tables I and III of the paper on the General Boiling Point Law in the May JOURNAL (pp. 732 and 735). In Table I, the first number should read 1.45 instead of 0.24. In Table III the numbers given in the last column should be stricken out. The corrected values may readily be computed from the data in the two preceding columns. The foregoing corrections do not affect the remainder of the paper in any essential respect.

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**Correction.**—In the recent paper by Keyes and Felsing<sup>1</sup> on “The Equation of State for Liquids and Vapors. I. The Vapor Phase of Ethyl Ether,” the following corrections should be made:

P. 594: Equation 5 should read

$$p = \frac{R_x T}{v - \delta_x} - \frac{A_x}{(v - l_x)^2}.$$

P. 597: 3rd paragraph, middle of page, 2nd sentence, replace “Equation 2” by “Equation 1,” and insert “scale” between “no” and “corrections” immediately following.

P. 597: footnote 3, omit the last sentence, beginning “Strictly . . . .”

P. 610: Table II. The columns should be headed, beginning with the second column

$$\delta, \delta, \frac{R}{v - \delta}, \frac{R}{v - \delta}, \phi, \phi.$$

P. 611: Third sentence on the page reading “The equation. . . .” change  $e$  to  $l$  and change corrections to correctness.

P. 611: In Table III change the number 7058.6 in the pressure equation to 7050.6

P. 618: In Table VI set  $2\beta_2 = 6.1748 \equiv 6.216 = \beta_1$  instead of  $2\beta_2 = 6.748 \equiv 6.216 = \beta_1$ .

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<sup>1</sup> THIS JOURNAL, 41, 589 (1919).